Deuterium content of natural water: an environmental risk factor for human disorders

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Budapest 4th International Congress on Deuterium Depletion October 17- 18, 2019

Definitions

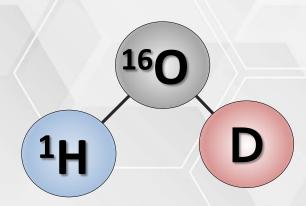
Isotopes are variants of a particular CHEMICAL ELEMENT with the same atomic number but with different atomic masses.

$$D = {}^{2}H$$

D = ²H= Heavy Hydrogen=Deuterium

- ¹H Hydrogen
- ¹⁶O Oxygen 16
- 17O Heavy Oxygen 17
- 18O Heavy Oxygen 18

Isotopologues are MOLECULES that differ only in their isotopic composition.

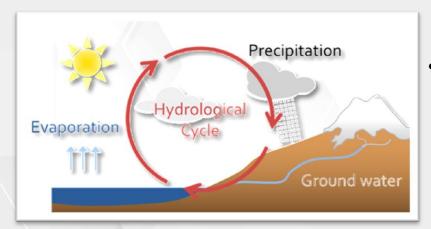


 $HOD = {}^{1}HD^{16}O = {}^{1}H^{2}H^{16}O = Semi-Heavy Water$

g/Kg WSMOW
997,032536356
2,227063738
0, 411509070
0, 328000097
0,000 728769
0,000 134998
0,0000 26900
0,0000000 59
0,000000011

Definitions

 DDW - Deuterium-depleted water is water obtained by depleting deuterium-containing isotopologues from natural water.



Low Deuterium Water is water with a lower concentration of deuterium than occurs naturally in certain area.

• "Light Water" is water with a higher concentration of the lightest stable isotopologue ¹H₂¹⁶O than occurs naturally in certain area.

Both natural-hydrological processes and technological deuterium depletion lead to an increase isotopologue ¹H₂¹⁶O and decrease HOD.



Deuterium generates a risk of depression



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Research report

Deuterium content of water increases depression susceptibility: The potential role of a serotonin-related mechanism

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Chronic stress study

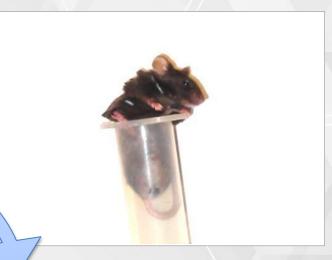
Exposure to rat



Resident-Intruder test «Social defeat»



Restraint stress



10 days Tail Suspension



Post-stress tests

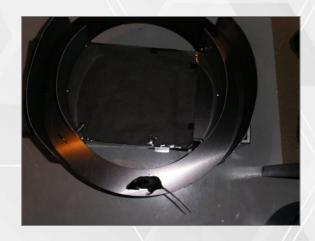
Sucrose preference test

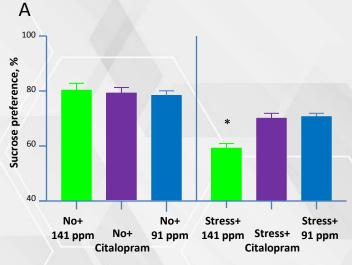


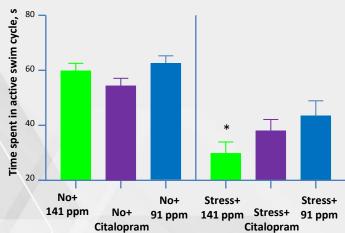
Forced swim test «Behavioral despair»

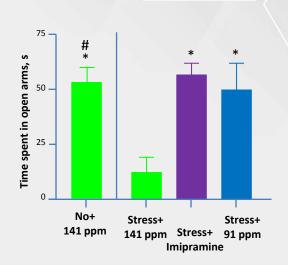


O-maze test



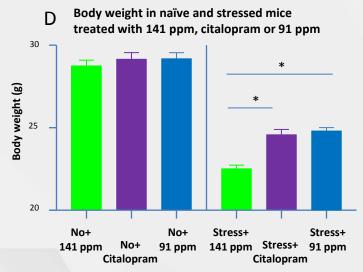




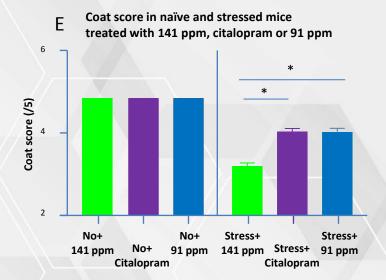


Post-stress tests and EEG sleep study

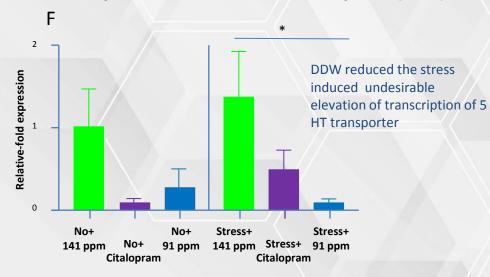
Body weight, g

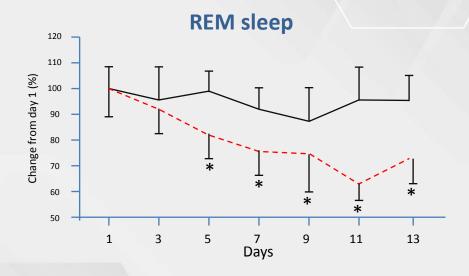


Coat state score, n

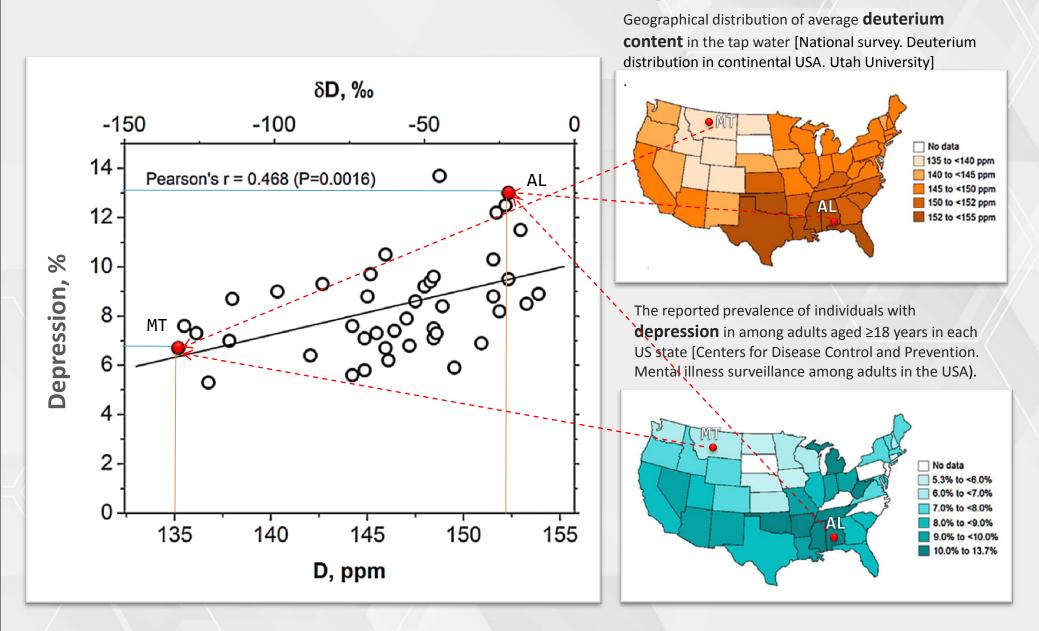


mRNA expression level of the 5-HT transporter (SERT)

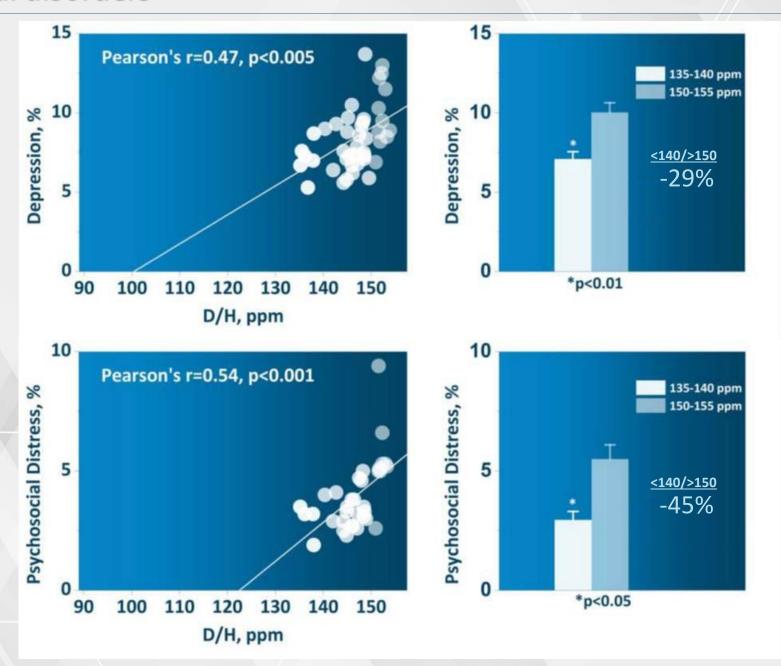




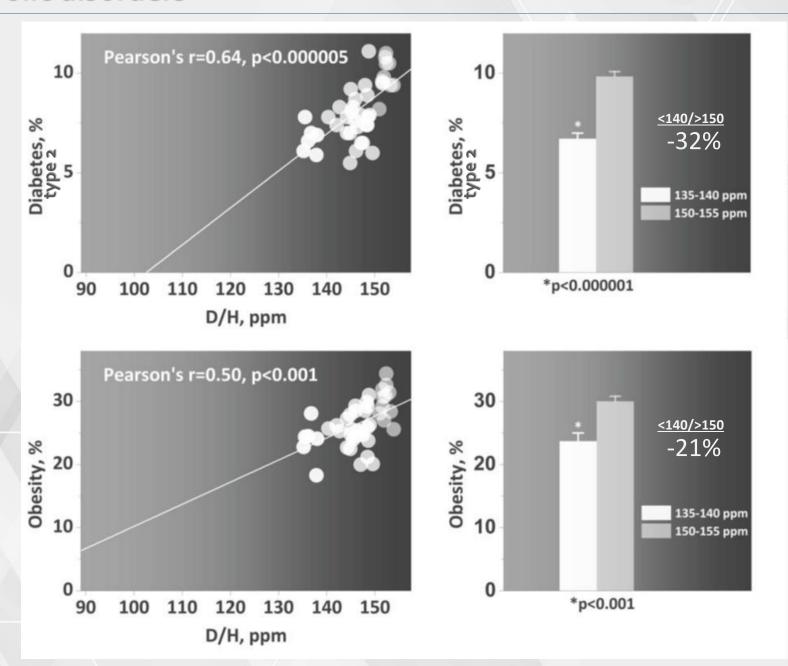
Correlation between reported rates of major depressive disorder in the US states and content of deuterium in tap water in the USA



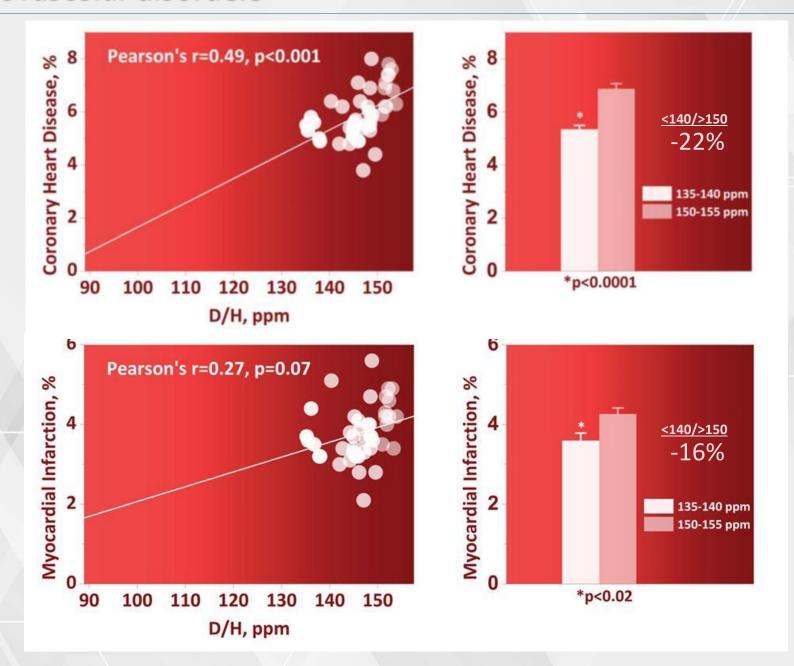
Mental disorders



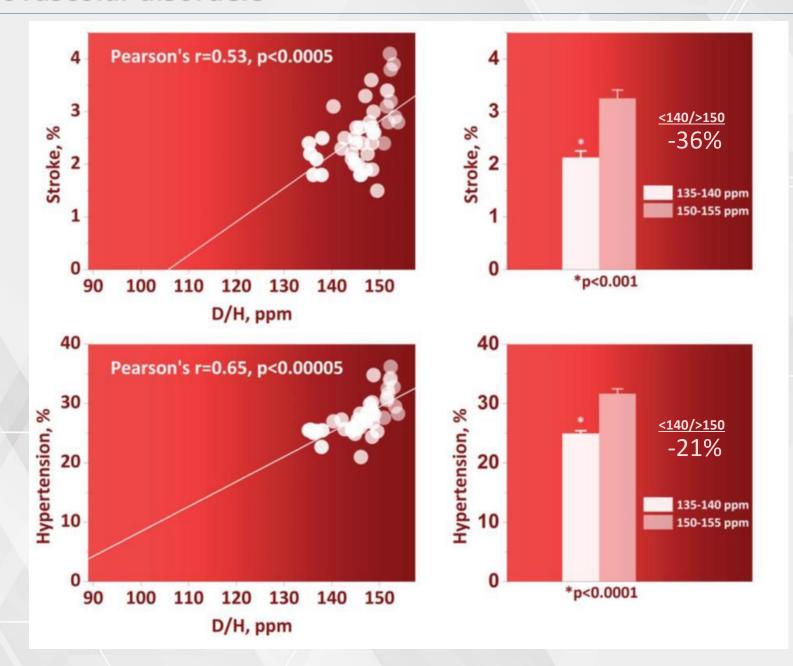
Metabolic disorders



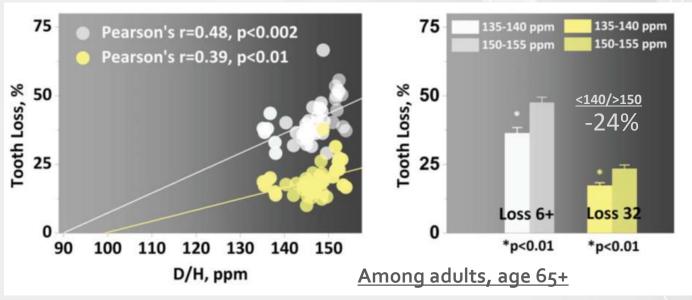
Cardiovascular disorders



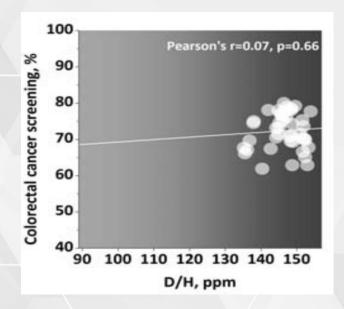
Cardiovascular disorders

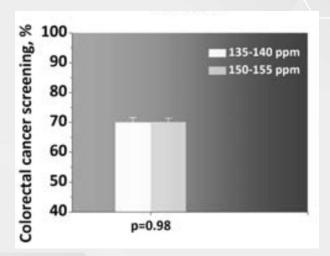


Oral health



Oncology





Summary

<u>Disorders</u>	Prevalence of disorders, < 140 ppm vs. >150 ppm					Correla	tion by states	
	Absolute <140 ppm	Absolute >150 ppm	Relative <140/>150	statistical significance p<	*	Pearson's coefficient	statistical significance p<	*
Psychosocial Distress	3.0%	5.5%	-45%	0.0005	*	0.54	0.001	*
Depression	7.1%	10.0%	-29%	0.01	*	0.47	0.005	*
Diabetes type 2	6.7%	9.8%	-32%	0.000001	*	0.64	0.000005	*
Obesity	23.7%	30.0%	-21%	0.001	*	0.50	0.001	*
Hypertension	25.0%	31.7%	-21%	0.0001	*	0.65	0.00005	*
Coronary Heart Disease	5.4%	6.9%	-22%	0.0001	*	0.49	0.001	*
Stroke	2.1%	3.3%	-36%	0.001	*	0.43	0.005	*
Myocardial Infarction	3.6%	4.3%	-16%	0.02	*	0.27	0.07	
Loss of 6 or more teeth	36.4%	47.6%	-24%	0.01	*	0.48	0.002	*

Replacement of plain water with deuterium depleted water may significantly reduce the risk of disorders

That's one small step in deuterium depletion, one giant leap for mankind's health.





Thank you!

Acknowledgements

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Supplemental material

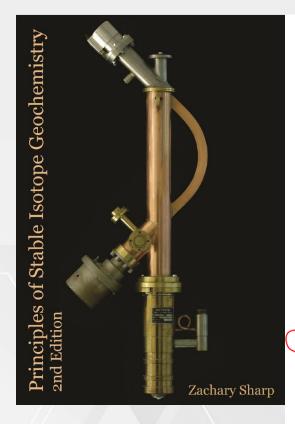


Table 2.1. Common Mistakes in Terminology and Phraseology

Mistake	Recommended Expressions	Explanation
referring to the symbol δ as 'del'	Since the time of the early Greeks, the name of this symbol has been and remains delta .	The word <i>del</i> describes either of two things in mathematics and science: an operator (∇) or the partial derivative (\hat{c})
δ ¹³ C composition	$\delta^{13}\text{C}$ value; carbon isotope composition	δ^{13} C values are numbers and a composition of numbers has no meaning.
Isotopically depleted water	¹⁸ O (or D) depleted water	A given sample of water is neither depleted nor enriched in isotopes.
Stable water isotopes have been widely used as tracers	Stable O and H isotope ratios in water have been used	Only elements have isotopes. It is the H and O that has the isotopes, not H ₂ O.
heavy (light) δ ¹⁸ O values	high (low) δ ¹⁸ O values	As numbers, δ -values can be high or low, positive or negative, but not heavy or light.
Isotopically negative	relatively low δ values	Isotopic ratios are not negative or positive; they are lower or higher than those of the standards.
depleted δ ¹³ C value	low δ^{13} C value (relative to another)	δ^{13} C values are numbers and, as such, they cannot be depleted or enriched.
enriched (depleted) carbonates.	isotopically heavy (light) carbonates	The words <i>enrich</i> and <i>deplete</i> are
enriched (depleted) compositions	(relatively) ¹⁸ O-rich or ¹³ C-poor carbonates	overused and much abused. These words should be reserved for describing a process that changes the
depleted carbon reservoir	reservoir of (isotopically) light carbon	content of the heavy isotope of the element in some substance.
oxygen isotopes in chert; inferred from carbon isotopes; isotopes of soil water	oxygen isotope ratio (composition) of chert; inferred from carbon isotope measurements; isotopic composition of soil water	Such written mistakes are a carryover from loose oral communication.